Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

## Shou-Cai Zhang, Guang-Bo Che* and Bo Liu

Department of Chemistry, Jilin Normal University, Siping 136000, People's Republic of China

Correspondence e-mail:
guangbochejl@yahoo.com

## Key indicators

Single-crystal X-ray study
$T=292 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.006 \AA$
$R$ factor $=0.043$
$w R$ factor $=0.096$
Data-to-parameter ratio $=15.0$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]
## 9-[4-(6-Bromo-4-phenyInaphthalen-2-yl)-phenyl]-9H-carbazole

The title compound, $\mathrm{C}_{34} \mathrm{H}_{22} \mathrm{BrN}$, was synthesized by N alkylation of 7-bromo-3-(4-bromophenyl)-1-phenylnaphthalene with $9 H$-carbazole. The carbazole ring system is essentially planar and makes a dihedral angle of 48.63 (9) ${ }^{\circ}$ with the plane of the adjacent benzene ring.

## Comment

A remarkable series of derivatives of carbazole has been investigated because of their diverse photophysical behavior (Adachi et al., 2001). Previously, the most quantitative attention has been focused on the hole-transporting (HT) properties for organic light-emitting devices (Li et al., 2005; Wong et al., 2005). In recent years, much interest has been paid to their crystal structures (Chen et al., 2005; Cui et al., 2006; Huang et al., 2005). In this research work, our idea is to synthesize a new derivative of carbazole with excellent HT ability and luminescent properties. Here, we report the crystal structure of the title compound, (I).

(I)

Selected bond lengths and angles for (I) are given in Table 1. The carbazole ring system is essentially planar and makes a dihedral angle of 48.63 (9) ${ }^{\circ}$ with the plane of the adjacent C13-C18 benzene ring. The dihedral angles formed by the C13-C18 and C29-C34 benzene rings with the naphthalene system are 16.15 (8) and $49.29(12)^{\circ}$, respectively (Fig. 1).

There are no supramolecular interactions, such as hydrogen bonds or $\pi-\pi$ stacking forces, in the crystal structure of (I).

## Experimental

Carbazole ( $1.0 \mathrm{mmol}, 167.2 \mathrm{mg}$ ) and sodium hydroxide ( 10.0 mmol , 400.0 mg ) were dissolved in dimethylformamide (DMF, 40 ml ). After stirring at room temperature for 30 min , a solution of 7-bromo-3-(4-bromophenyl)-1-phenylnaphthalene ( $437.8 \mathrm{mg}, 1.0 \mathrm{mmol}$ ) in DMF was added dropwise with stirring. The mixture was refluxed with

Received 11 March 2006
Accepted 4 April 2006
stirring for 3.5 h to obtain a clear yellow solution, which was then distilled under reduced pressure. The yellow solid residue was purified by column chromatography on silica gel using petrol ether/ethyl acetate $(10: 3 \mathrm{v} / \mathrm{v})$ as eluent. Fine colourless single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of the eluent at 298 K (yield 90\%).

## Crystal data

$$
\begin{aligned}
& \mathrm{C}_{34} \mathrm{H}_{22} \mathrm{BrN} \\
& M_{r}=524.44 \\
& \text { Orthorhombic, Pbcn } \\
& a=15.9337(11) \AA \\
& b=10.8421(8) \AA \\
& c=28.542(2) \AA \\
& V=4930.7(6) \AA
\end{aligned}
$$

$$
\begin{aligned}
& Z=8 \\
& D_{x}=1.413 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo Ka radiation }
\end{aligned}
$$

$$
\begin{aligned}
& \text { Mo } K \alpha \text { radiation } \\
& \mu=1.69 \mathrm{~mm}^{-1}
\end{aligned}
$$

$T=292$ (2) K
Block, colourless
$0.40 \times 0.21 \times 0.12 \mathrm{~mm}$

## Data collection

Bruker SMART CCD area-detector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan (SADABS; Bruker, 1998)
$T_{\text {min }}=0.656, T_{\text {max }}=0.815$

## Refinement

Refinement on $F^{2}$
H-atom parameters constrained
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.043$
$w R\left(F^{2}\right)=0.096$
$S=0.78$
4871 reflections
325 parameters
$w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}{ }^{2}\right)+(0.0375 P)^{2}\right]$
where $P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3$
$(\Delta / \sigma)_{\max }=0.001$ 。
$\Delta \rho_{\max }=0.42 \mathrm{e}^{\circ}{ }^{-3}$
$\Delta \rho_{\max }=0.42 \mathrm{e}^{2} \AA^{-3}$
$\Delta \rho_{\min }=-0.42 \mathrm{e}^{-3}$

Table 1
Selected geometric parameters ( $\left({ }^{\circ},{ }^{\circ}\right)$.

| $\mathrm{C} 9-\mathrm{N}$ | $1.401(4)$ | $\mathrm{C} 13-\mathrm{N}$ | $1.421(4)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{C} 12-\mathrm{N}$ | $1.397(4)$ | $\mathrm{C} 23-\mathrm{Br}$ | $1.893(3)$ |
|  |  |  |  |
| $\mathrm{C} 14-\mathrm{C} 13-\mathrm{N}$ | $118.7(3)$ | $\mathrm{C} 12-\mathrm{N}-\mathrm{C} 13$ | $125.1(3)$ |
| $\mathrm{C} 18-\mathrm{C} 13-\mathrm{N}$ | $121.5(3)$ | $\mathrm{C} 9-\mathrm{N}-\mathrm{C} 13$ | $126.6(3)$ |

All H atoms were positioned geometrically and refined as riding, with $\mathrm{C}-\mathrm{H}=0.93 \AA$ and $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C})$.


Figure 1
The molecular structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the $50 \%$ probability level. H atoms have been omitted for clarity.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1998); software used to prepare material for publication: SHELXTL.

The authors thank Professor Ning-Hai Hu and Heng-Qing Jia of Changchun Institute of Applied Chemistry, Chinese Academy of Sciences, for supporting this work.

## References

Adachi, C., Baldo, M. A., Forrest, S. R., Lamansky, S., Thompson, M. E. \& Kwomg, R. C. (2001). Appl. Phys. Lett. 78, 1622-1624.
Bruker (1998). SMART, SAINT, SADABS and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
Chen, L.-Q., Yang, C.-L., Meng, X.-G. \& Qin, J.-G. (2005). Acta Cryst. E61, o3073-o3075.
Cui, J.-L., Huang, P.-M. \& Guo, W.-L. (2006). Acta Cryst. E62, o143-o144.
Huang, P.-M., Li, J.-S., Duan, X.-M., Zeng, T. \& Yan, X.-L. (2005). Acta Cryst. E61, o2366-o2367.
Li, J., Liu, D., Li, Y., Lee, C.-S., Kwong, H.-L. \& Lee, S. (2005). Chem. Mater. 17, 1208-1212.
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany
Wong, K.-T., Chen, Y.-M., Lin, Y.-T., Su, H.-C. \& Wu, C.-C. (2005). Org. Lett. 7, 5361-5364.


[^0]:    (C) 2006 International Union of Crystallography All rights reserved

